

## ALKALOIDS FROM THE LEAVES OF *BUXUS SEMPERVIRENS*

ATTA-UR-RAHMAN,\* DILDAR AHMED, M. IQBAL CHOUDHARY,

H.E.J. Research Institute of Chemistry, University of Karachi, Karachi 32, Pakistan

BILGE SENER, and SONGUL TURKOZ

Department of Pharmacognosy, Faculty of Pharmacy, Gazi University, Ankara, Turkey

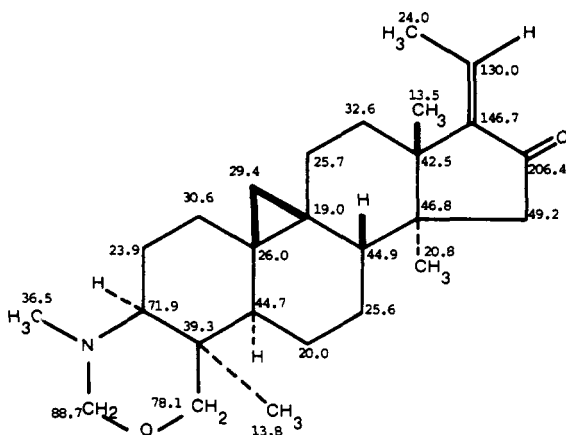
*Buxus sempervirens* L. (Buxaceae) is a shrub that occurs extensively in Eurasia. *Buxus* species extracts have been used in indigenous medicine for a variety of purposes, including as a febrifuge and for the relief of rheumatism and malaria (1). Previous work on *Buxus* species has resulted in the isolation of over 135 steroidal alkaloids. In continuation of our studies on various *Buxus* species (2-11), we now report the isolation and structure of a new alkaloid, (-)-sempervirone [1]. Two other bases isolated afforded spectral data consistent with the structures corresponding to (-)-cyclomicrobuxinine [2] (12) and (-)-cyclosuffrobuxinine M [3] (14). All these compounds have been isolated from the so-called weakly basic alkaloidal fraction of *B. sempervirens* leaves, extractable between pH 3.0 and 5.0.

(-)-Sempervirone [1],  $C_{26}H_{39}NO_2$ ,  $[\alpha]^{20}_D -65$  ( $CHCl_3$ ) showed a uv maximum at 243 nm, characteristic of an  $\alpha,\beta$ -unsaturated ketone. The ir spectrum exhibited strong absorptions at 1715 and  $1632\text{ cm}^{-1}$  ( $\alpha,\beta$ -unsaturated cyclopentanone) (14).

The  $^1H$ -nmr spectrum (400 MHz,  $CDCl_3$ ) displayed three 3-H singlets at  $\delta$  0.94, 1.16, and 1.32, corresponding to Me-32, Me-30, and Me-18, respectively. An AB quartet centered at  $\delta$  0.43 and 0.72 ( $J_{19\alpha,19\beta} = 4.5$  Hz) was assigned to  $\alpha$ - and  $\beta$ -cyclopropyl methylene protons. A 3-H doublet at  $\delta$  1.83 ( $J = 7.5$  Hz) was due to Me-21 coupled with the vinylic H-20, which in turn appeared as a quartet at  $\delta$  6.55 ( $J = 7.5$  Hz) (15). Additionally, a 3-H singlet at  $\delta$  2.16 was attributed to the *N*-Me group. An AB quartet resonating at  $\delta$  3.16 and 3.77

( $J_{31\alpha,31\beta} = 10.6$  Hz) was assigned to the C-31 methylenic protons, while another AB quartet at  $\delta$  3.63 and 4.46 ( $J_{33\alpha,33\beta} = 7.5$  Hz) was due to the C-33 methylenic protons of the tetrahyooxazine ring (3).

Two-dimensional  $^1H$ -nmr measurements (COSY  $45^\circ$ , NOESY) (16) fully agreed with the proposed structure 1 for (-)-sempervirone. The COSY- $45^\circ$  spectrum established the coupling interactions between the geminal and vicinal protons. Thus, the COSY spectrum showed geminal interactions between C-19 $\alpha$  and  $\beta$  protons as well as between C-31 $\alpha$  and  $\beta$  methylenic protons. The NOESY spectrum served to show the relative stereochemistry at several key points in the molecule. Strong NOESY cross peaks were observed between the 18-methyl and 21-allylic methyl protons. This observation supported an *E* geometry for the enone system. The  $^{13}C$ -nmr spectrum ( $CDCl_3$ , 100 MHz) of (-)-sempervirone [1] showed four signals at  $\delta$  13.45, 13.79, 20.82, and 23.96 which have been assigned to the methyl carbons C-18, C-30, C-32, and C-21, respectively. The C-31 methylene carbon appeared at  $\delta$  78.07, and the C-33 methylene carbon resonated at  $\delta$  88.66 due to the deshielding influence of the nitrogen and oxygen atoms, respectively. The assignments to the various carbons, confirmed by DEPT and GASPE experiments, are presented around structure 1. The mass spectrum of compound 1 included the molecular ion at  $m/z$  397.2982, in agreement with the molecular formula  $C_{26}H_{39}NO_2$  (calcd 397.2981). The peak at  $m/z$  382.2748 resulted from the loss of a



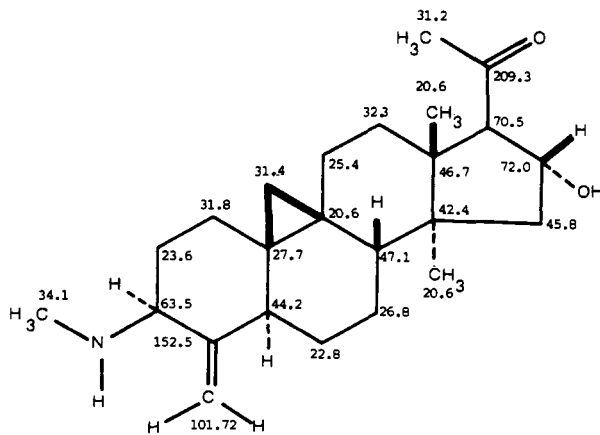
1

methyl group from the molecular ion. A considerably larger peak at  $m/z$  127.1002 corresponded to the cleavage of ring A along with the attached tetrahydrooxazine ring. This fragment is common in compounds bearing a non-substituted tetrahydrooxazine ring (3). The base peak at  $m/z$  71.0496, having the formula  $C_4H_9N$  (calcd 71.0496), was due to the cleavage of ring A along with a part of the nitrogen-containing side ring, accompanied by an intramolecular proton transfer (3). These data led to structure **1** for (-)-sempervirone.

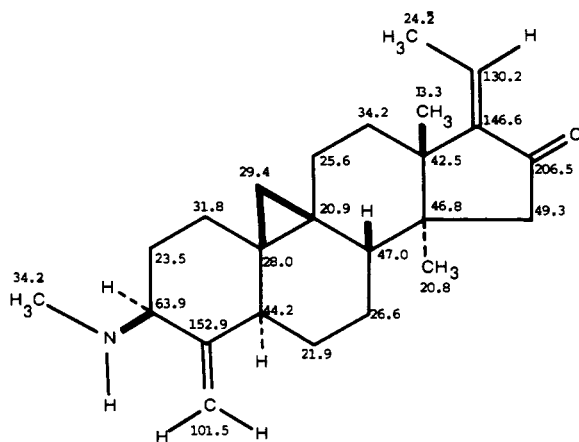
Our second compound was identified

to have the structure corresponding to (+)-cyclomicrobuxinine [**2**] by comparison of its spectral data (ms, nmr, uv, and ir) with those reported in the literature (12,13) and by the consistency of the spectral data with structure **2**. Presently the  $^{13}C$ -nmr assignments, confirmed by DEPT experiments (17), are presented around structure **2**.

Our third base was identified as (-)-cyclosuffrobuxinine M [**3**], previously isolated from *Buxus microphylla* (14), by spectral assignments, and by comparison of the spectral data with those reported in the literature. We have ob-



2



3

tained it for the first time from the leaves of *B. sempervirens*. Its  $^{13}\text{C}$ -nmr assignments are presented around structure **3**.

### EXPERIMENTAL

**PLANT MATERIAL.**—The leaves of *B. sempervirens* (dry wt 10 kg) were collected from the Beynam Forest, Ankara, Turkey, in September 1986. The plant was identified by Dr. Bilge Sener, Department of Pharmacognosy, Gazi University, and a voucher specimen was deposited in the herbarium of the Faculty of Pharmacy, Gazi University, Ankara.

**EXTRACTION AND PURIFICATION.**—The EtOH extract of air-dried leaves was evaporated to a gum. The total alkaloids (500 g) were obtained by extraction into 10% HOAc. Partial separation of the alkaloids was achieved by extraction into  $\text{CHCl}_3$  at different pH values. The fraction obtained at pH 3.5 (20 g) was loaded on a Si gel column (300 g). Elution was carried out with  $\text{CHCl}_3$  and then with  $\text{CHCl}_3/\text{MeOH}$ . Three main fractions were obtained: Fraction A,  $\text{CHCl}_3/\text{MeOH}$  (98:2); fraction B,  $\text{CHCl}_3/\text{MeOH}$  (90:10); fraction C,  $\text{CHCl}_3/\text{MeOH}$  (85:15).

**(-)-SEMPERVIRONE [1].**—Fraction A was further purified by preparative tlc (Si gel) in  $\text{C}_6\text{H}_{14}-\text{Me}_2\text{CO}$  (2:1) to afford (-)-sempervirone [1] as a white amorphous solid (6.0 mg),  $[\alpha]_D^{20} -65^\circ$  ( $c = 0.5$ ,  $\text{CHCl}_3$ );  $\lambda$  max (MeOH) 243 nm ( $\log \epsilon$  3.8);  $\nu$  max ( $\text{CHCl}_3$ ) 1715, 1638 ( $\alpha, \beta$ -unsaturated ketone/cyclopentenone), 1150 (C-O)  $\text{cm}^{-1}$ ;  $^1\text{H}$  nmr (400 MHz,  $\text{CDCl}_3$ )  $\delta$  0.43 (1H, d,  $J_{19\alpha, 19\beta} = 4.5$  Hz, H-19 $\alpha$ ), 0.72 (1H, d,  $J_{19\beta, 19\alpha} = 4.5$  Hz, H-19 $\beta$ ), 0.94 (3H, s, Me-32), 1.16 (3H, s, Me-30), 1.32 (3H, s, Me-18), 1.83 (3H, d,  $J_{21, 20} = 7.5$  Hz, Me-21), 2.16 (3H, s, N-Me), 3.16 (1H, d,  $J_{33\alpha, 33\beta} = 10.6$  Hz, H-33), 3.63 (1H, d,  $J_{31\alpha, 31\beta} = 7.5$  Hz, H-31), 3.77 (1H, d,  $J_{33\beta, 33\alpha} = 10.6$  Hz, H-33), 4.46

(1H, d,  $J_{31\beta, 31\alpha} = 7.5$  Hz, H-31), 6.55 (1H, q,  $J = 7.5$  Hz, H-20);  $m/z$  (rel. int.)  $[\text{M}]^+$   $\text{C}_{26}\text{H}_{39}\text{NO}_2$  397.2980 (31),  $\text{C}_{25}\text{H}_{36}\text{NO}_2$  382.7441 (27),  $\text{C}_7\text{H}_{13}\text{NO}$  127.1002 (45),  $\text{C}_5\text{H}_{11}\text{N}$  85.0699 (65),  $\text{C}_4\text{H}_{10}$  72.1311 (29),  $\text{C}_4\text{H}_9\text{N}$  71.0859 (100),  $\text{C}_4\text{H}_8\text{N}$  70.0658 (53),  $\text{C}_3\text{H}_8\text{N}$  58.0579 (90).

**(+)-CYCLOMICROBUXININE [2].**—Fraction B was further chromatographed by preparative tlc (Si gel) in  $\text{C}_6\text{H}_{14}-\text{Me}_2\text{CO}-\text{Et}_2\text{NH}$  (7:3:2) to afford a colorless, amorphous solid **2** (20 mg),  $[\alpha]_D^{20} +162$  ( $c = 1.5$ ,  $\text{CHCl}_3$ );  $\lambda$  max (MeOH) terminal;  $\nu$  max ( $\text{CHCl}_3$ ) 3600 (N-H), 3390 (O-H), 1695 (ketone carbonyl)  $\text{cm}^{-1}$ ;  $^1\text{H}$  nmr (300 MHz,  $\text{CDCl}_3$ )  $\delta$  0.06 (1H, d,  $J_{19\alpha, 19\beta} = 4.6$  Hz, H-19 $\alpha$ ), 0.30 (1H, d,  $J_{19\beta, 19\alpha} = 4.6$  Hz, H-19 $\beta$ ), 0.90 (3H, s, Me), 1.20 (3H, s, Me), 2.14 (3H, s, Me), 2.49 (3H, s, N-Me), 2.89 (1H, dd,  $J_{3,2} = 11.8$  Hz,  $J_{3,2} = 4.4$  Hz, H-3), 3.01 (1H, d,  $J_{17, 16} = 6.6$  Hz, H-17), 4.58 (1H, s, =C-H), 4.85 (1H, s, =C-H), 4.89 (1H, ddd,  $J_{16, 17} = 6.6$  Hz,  $J_{16, 15} = 7.6$  Hz,  $J_{16, 15} = 2.0$  Hz, H-16);  $^{13}\text{C}$  nmr (100 MHz,  $\text{CDCl}_3$ ) see structure **2**;  $m/z$  (rel. int.)  $[\text{M}]^+$   $\text{C}_{24}\text{H}_{37}\text{NO}_2$  371.2831 (calcd 371.2824) (19),  $[\text{M} - \text{Me}]^+$  356 (95),  $[\text{M} - \text{Ac}]^+$  328 (13),  $\text{C}_3\text{H}_7\text{N}$  57 (100).

**(-)-CYCLOSUFFROBUXININE M [3].**—Fraction C was purified further by tlc (Si gel) in solvent system  $\text{C}_6\text{H}_{14}-\text{Me}_2\text{CO}-\text{Et}_2\text{NH}$  (6:4:2) to afford **3** (10 mg),  $[\alpha]_D^{20} -56$  ( $c = 1.8$ ,  $\text{CHCl}_3$ );  $\lambda$  max (MeOH) 243 nm ( $\log \epsilon$  3.2);  $\nu$  max ( $\text{CHCl}_3$ ) 3610 (NH), 1705, 1640 ( $\alpha, \beta$ -unsaturated ketone/cyclopentenone);  $^1\text{H}$  nmr (400 MHz,  $\text{CDCl}_3$ )  $\delta$  0.13 (1H, d,  $J_{19\alpha, 19\beta} = 4.5$  Hz, H-19 $\alpha$ ), 0.38 (1H, d,  $J_{19\beta, 19\alpha} = 4.5$  Hz, H-19 $\beta$ ), 0.97 (3H, s, Me), 1.32 (3H, s, Me), 1.82 (3H, d,  $J_{21, 20} = 7.5$  Hz, Me-21), 2.54 (3H, s, N-Me), 2.97 (1H, br d,  $J_{3,2} = 10.4$  Hz, H-3), 4.84 (1H, s, =C-H), 4.86 (1H, s, =C-H), 6.57 (1H, q, H-20);  $^{13}\text{C}$  nmr (100 MHz,  $\text{CDCl}_3$ ) see structure **3**;  $m/z$

(rel. int.)  $[M]^+$   $C_{24}H_{35}NO$  353.2725 (calcd 353.2718),  $[M - Me]^+$  338 (8),  $C_4H_9N$  71 (60),  $C_3H_8N$  58 (100).

#### ACKNOWLEDGMENTS

The authors wish to express their thanks to Glaxo Laboratories, Ltd., Pakistan for financial support for D.A.

#### LITERATURE CITED

1. G.A. Cordell, "Introduction to Alkaloids," Wiley-Interscience, New York, 1981, p. 907.
2. Atta-ur-Rahman, M. Nisa, and T. Zamir, *Z. Naturforsch.*, **39b**, 127 (1984).
3. Atta-ur-Rahman, M.I. Choudhary, and M. Nisa, *Heterocycles*, **23**, 1951 (1985).
4. Atta-ur-Rahman, M.I. Choudhary, I. Ali, and Habib-ur-Rehman, *J. Nat. Prod.*, **49**, 106 (1986).
5. M.I. Choudhary, Atta-ur-Rahman, A.J. Freyer, and M. Shamma, *Tetrahedron*, **42**, 5747 (1986).
6. M.I. Choudhary, Atta-ur-Rahman, A.J. Freyer, and M. Shamma, *J. Nat. Prod.*, **50**, 84 (1987).
7. Atta-ur-Rahman, M. Nisa, T. Zamir, and W. Voelter, *Z. Naturforsch.*, **40b**, 565 (1985).
8. Atta-ur-Rahman, M. Nisa, and S. Farhi, *Z. Naturforsch.*, **39b**, 524 (1984).
9. Atta-ur-Rahman and M. Nisa, *Z. Naturforsch.*, **39b**, 839 (1984).
10. Atta-ur-Rahman, M. Nisa, and K. Jahan, *Phytochemistry*, **24**, 1398 (1985).
11. Atta-ur-Rahman, M.I. Choudhary, and M. Nisa, *Planta Med.*, **75** (1987).
12. T. Nakano and M. Hasegawa, *J. Chem. Soc. C*, 6688 (1965).
13. Z. Voticky, J. Tomko, L. Dolejs, and V. Hanus, *Collect. Czech. Chem. Commun.*, **30**, 3705 (1965).
14. T. Nakano, S. Terao, and Y. Saeki, *J. Chem. Soc. C*, 1412 (1966).
15. K.S. Brown, Jr., and S.M. Kupchan, *J. Am. Chem. Soc.*, **86**, 4414 (1964).
16. Atta-ur-Rahman, "Nuclear Magnetic Resonance Spectroscopy," Springer-Verlag, New York, 1986, p. 202.
17. G.A. Morris, *Magn. Res. Chem.*, **24**, 371 (1986).

Received 19 October 1987